Tetrahedron Letters No.38, pp. 2721-2724, 1964. Pergamon Press Ltd. Printed in Great Britain.

1,3,2-Dioxaborinium and 4,5-benzo-1,3,2-dioxaborinium cations, novel aromatic systems

A. T. Balaban, E. Barabas and A. Arsene Institute of Atomic Physics, P. O. Box 35, Bucharest, Romania (Received 24 July 1964)

In a previous study of the reaction between 2-alkoxy-1,3,2-dioxaboroles and bidentate ligands (1), it was shown that with enolizable 1,3-diketones or benzoderivatives such as \underline{o} -hydroxyacetophenone, crystalline amphionic spiroborates I and II are formed.



No crystalline products III and IV could be obtained from bidentate ligands and tri-n-butyl borate. However, on treating an equimolar mixture of tri-n-butyl borate and 1,3diketone with a solution of anhydrous perchloric acid in methylene dichloride (2), crystalline perchlorates are formed (with aliphatic diketones, petroleum ether must be added).

2721

1,3,2-Dioxoaborinium and 4,5-benzo-1,3,2-dioxaborinium cations

Elemental analyses and the mode of formation indicate for these compounds formulas V and VI, respectively; in their infrared spectra, bands due to the perchlorate anion (629 and 1100 cm⁻¹) are always present.





Thus acetylacetone affords V (R = R" = Me, R' = E), m.p. 176° dec. (from dioxane - petroleum ether) ; UV bands in methylene dichloride 252 sh and 293 mµ ; in acetonitrile 225, 249 sh and 292 mµ ; IR bands in nujol : 840, 900, 928, 970, 1007, 1025, 1570 broad, 3085 and 3145 cm⁻¹. 3-kethyl-2,4-pentanedione affords V (R = R' = R" = Me), m.p. 175° dec. (from acetonitrile) ; UV bands in methylene dichloride 264 sh and 313 mµ ; in acetonitrile 232, 264 sh and 311 mµ ; IR bands in nujol : 888, 935, 1002, 1211, 1335, 1525 broad and 1575 cm⁻¹. Dibenzoylmethane in benzene affords V (R = = R" = Ph, R' = H), yellow, which may be recrystallized from acetonitrile and explodes without melting ; an excess of perchloric acid leads to a by-product with m.p. 260° which was not further investigated ; UV bands in methylene dichlo-

2722

No.38

1,3,2-Dioxaborinium and 4,5-benzo-1,3,2-dioxaborinium cations

ride 250 sh, 277, 310. 382 and 400 mµ ; in acetonitrile 239, 250 sh, 275, 307, 379 and 397 mµ ; IR bands in nujol and in potassium bromide pellet : 610, 675, 719, 728, 789, 910, 1002, 1041, 1171, 1193, 1268, 1313, 1360, 1385, 1439, 1494, 1520, 1560 broad, 1606, 3080 and 3150 cm⁻¹. <u>o</u>-Hydroxyacetophenone affords VI (R = Me), yellow, m.p. 223⁰ (from acetonitrile) ; UV maxima in methylene dichloride 254, 295, 330 and 387 mµ ; in acetonitrile 252, 290, 327 and 380 mµ ; the compound is very hygroscopic and the bands at 252-4 and 327-30 mµ due to <u>o</u>-hydroxyacetophenone formed by hydrolysis gradually increase ; prominent IR bands in nujol : 730, 771, 812, 1005, 1023, 1150, 1178, 1208, 1258, 1290, 1408, 1495, 1520 broad, 1586 and 1629 cm⁻¹. The IR spectral data evidence that in these systems there is strong chelation of the carbonyl groups and there is no free enol group, in agreement with formulas V and VI.



These compounds are derivatives of the 1,3,2-dioxaborinium cation VII for which several limiting structures may be written. Structure VIIa and the two structures VIIb evidence that this system can be regarded as a boroxaropyrylium cation (3), while the two structures VIIc denote that positions 4 and 6 must be strongly electrophilic. At the same time, this new aromatic system may be regarded as a Z-homologue (4) of 1,3,2-dioxaboroles.

2723

Reactions and further derivatives of these systems are under investigation and will be reported in <u>Revue Rou-</u> <u>maine de Chimie</u>.

References

- 1. A. T. Balaban, C. N. Rențea and M. Mocanu, <u>Tetrahedron</u> Letters (under press).
- 2. F. Klages and P. Hegenberg, Angew. Chem. 74, 902 (1962).
- 3. M. J. S. Dewar and R. Dietz, <u>J. Chem. Soc. 1959</u>, 2728; cf. P. M. Maitlis, <u>Chem. Rev. 62</u>, 223 (1962).
- 4. A. T. Balaban and Z. Simon, Tetrahedron, 18, 315 (1962).